

Fig.2. Pictorial representation of $\mathrm{Th}_{3} \mathrm{~N}_{4}$.

N atom has 3 Th neighbors at $2.31 \AA$ and 1 Th at $2.47 \AA$, while an octahedral N atom has 3 Th at $2.53 \AA$ and 3 Th at $2.91 \AA$. The unit cell is displayed in pictorial form in Fig. 2. Interatomic distances are listed in Table 2.

Table 2. Interatomic distances

| $\mathrm{Th}(\mathrm{I})-\mathrm{Th}(\mathrm{I})(6)$ | $3.875(1) \AA$ |
| :---: | :--- |
| $\mathrm{Th}(\mathrm{II})(6)$ | $3.780(18)$ |
| $\mathrm{N}(\mathrm{II})(6)$ | $2.532(13)$ |
| $\mathrm{Th}(\mathrm{II})-\mathrm{Th}(\mathrm{I})(3)$ | $3.780(18)$ |
| $\mathrm{Th}(\mathrm{II})(3)$ | $3.772(35)$ |
| $\mathrm{Th}(\mathrm{II})(6)$ | $3.775(1)$ |
| $\mathrm{N}(\mathrm{I})(3)$ | $2.308(8)$ |
| $\mathrm{N}(\mathrm{I})(1)$ | $2.468(31)$ |
| $\mathrm{N}(\mathrm{II})(3)$ | $2.910(22)$ |
| $\mathrm{N}(\mathrm{I})-\mathrm{N}(\mathrm{I})(3)$ | $2.935(28)$ |
| $\mathrm{N}(\mathrm{II})(3)$ | $3.303(26)$ |
| $\mathrm{N}(\mathrm{II})-\mathrm{N}(\mathrm{II})(3)$ | $3.259(40)$ |

## Rcferences

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The crystal structure of ammonium tris(pentasulfido)platinum(IV) dihydrate. A correction. By Philip E. Jones and Lewis Katz, Department of Chemistry and Institute of Materials Science, University of Connecticut, Storrs, Connecticut, U. S. A.
(Received 27 March 1970)
A corrected version of a figure is given.
Fig. 2 of a recent article (Jones \& Katz, 1969) was published incorrectly. Although the authors disclaim responsibility, they extend their sympathy to any reader who became dizzy trying to view this Figure.

## Reference

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Fig.2. Stereoscopic illustration of the unit cell and contents. The large circles indicate the positions of the water molecules and ammonium ions, which form chains weaving around the $\operatorname{Pt}\left(\mathrm{S}_{5}\right)_{3^{2-}}$ anions in the $z$ direction.

